

NACE

Reference to the Peabody Book  
Peabody's CONTROL OF PIPELINE CORROSION, 2nd Edition  
January 27th, 2007  
Revision 1.0  
US Standard Units

Ohm's Law

$$E = IR$$

$$I = \frac{E}{R}$$

$$R = \frac{E}{I}$$



**SAMPLE**

Current (amps) =  100.00 A  
Resistance (ohms) =  100.00 Ω  
Voltage (volts) =  100.00 V

$E =$   1000.00 V  
 $R =$   100.00 Ω  
 $I =$   10.00 A



$E$  = Voltage ( volts )  
 $I$  = Current ( amps )  
 $R$  = Resistance ( ohms )

$$R = \frac{E}{I}$$

$E =$   1000.00 V  
 $I =$   10.00 A  
 $R =$   100.00 Ω

### Area of a Circle

$$A = \pi R^2 = \pi \left( \frac{D}{2} \right)^2 = \frac{\pi D^2}{4}$$

$D$  = Diameter of Circle ( in )

$R$  = Radius of Circle ( in )

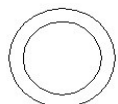
$A$  = Area of Circle ( in<sup>2</sup> )

$D$  = 10.00 in

$R$  = 5.00 in

$A$  = 78.54 in<sup>2</sup>

### Area of Pipe Cross-Section



$$A = \pi R_o^2 - \pi R_i^2 = \pi \left( \frac{D_o}{2} \right)^2 - \pi \left( \frac{D_i}{2} \right)^2$$

$D_o$  = Outer Diameter ( in )

$D_i$  = Inner Diameter ( in )

$R_o$  = Outer Radius ( in )

$R_i$  = Inner Radius ( in )

$A$  = Area of Pipe Cross-Section ( in<sup>2</sup> )

10.00 in  
5.00 in  
4.88 in  
3.88 in

Area of Pipe Cross-Section ( in<sup>2</sup> )

$S = 2 \pi R L = \pi D L$

$D$  = Diameter of Circle ( in )

$L$  = Length of Pipe ( in )

$A$  = Area of Pipe Cross-Section ( in<sup>2</sup> )

which is

10.00 in

300.00 in

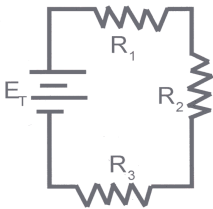
$S$  = 9424.78 in<sup>2</sup>

$S$  = 0.9425 ft<sup>2</sup>

Power	Actual Number	Abbrev.	Name
10 <sup>-24</sup>	0.000,000,000,000,000,000,000,1	y	yocto
10 <sup>-21</sup>	0.000,000,000,000,000,000,000,1	z	zepto
10 <sup>-18</sup>	0.000,000,000,000,000,001	a	atto
10 <sup>-15</sup>	0.000,000,000,000,001	f	femto
10 <sup>-12</sup>	0.000,000,000,001	p	pico
10 <sup>-9</sup>	0.000,000,001	n	nano
10 <sup>-6</sup>	0.000,001	μ	micro
10 <sup>-3</sup>	0.001	m	milli
10 <sup>-2</sup>	0.01	c	centi
10 <sup>-1</sup>	0.1	d	deci
10 <sup>0</sup>	1	---	---
10 <sup>1</sup>	10	D	deka
10 <sup>2</sup>	100	h or H	hecto
10 <sup>3</sup>	1,000	k or K	kilo
10 <sup>6</sup>	1,000,000	M	mega
10 <sup>9</sup>	1,000,000,000	G	giga
10 <sup>12</sup>	1,000,000,000,000	T	tera
10 <sup>15</sup>	1,000,000,000,000,000	P	peta
10 <sup>18</sup>	1,000,000,000,000,000,000	E	exa
10 <sup>21</sup>	1,000,000,000,000,000,000,000	Z	zeta

Series Circuit

$R_T = R_1 + R_2 + R_3$



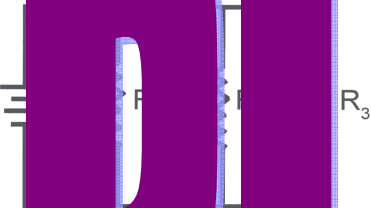
$R_1$  = Resistance of Resistor # 1 ( ohms )  
 $R_2$  = Resistance of Resistor # 2 ( ohms )  
 $R_3$  = Resistance of Resistor # 3 ( ohms )  
 $R_T$  = Total Resistance ( ohms )

$R_1$	=	100.00 $\Omega$
$R_2$	=	200.00 $\Omega$
$R_3$	=	0.00 $\Omega$
$R_T$	=	300.00 $\Omega$

Parallel Circuit

$$\frac{1}{R_T} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3}$$

$R_1$  = Resistance of Resistor # 1 ( ohms )  
 $R_2$  = Resistance of Resistor # 2 ( ohms )  
 $R_3$  = Resistance of Resistor # 3 ( ohms )  
 $R_T$  = Total Resistance ( ohms )



$R_1$	=	100.00 $\Omega$
$R_2$	=	200.00 $\Omega$
$R_3$	=	0.00 $\Omega$
$R_T$	=	66.67 $\Omega$

Parallel Circuit

$I_T = I_1 + I_2 + I_3$

$I_T$  = Total Current ( amps )  
 $R$  = Resistance ( ohms )  
 $P$  = Power ( watts )

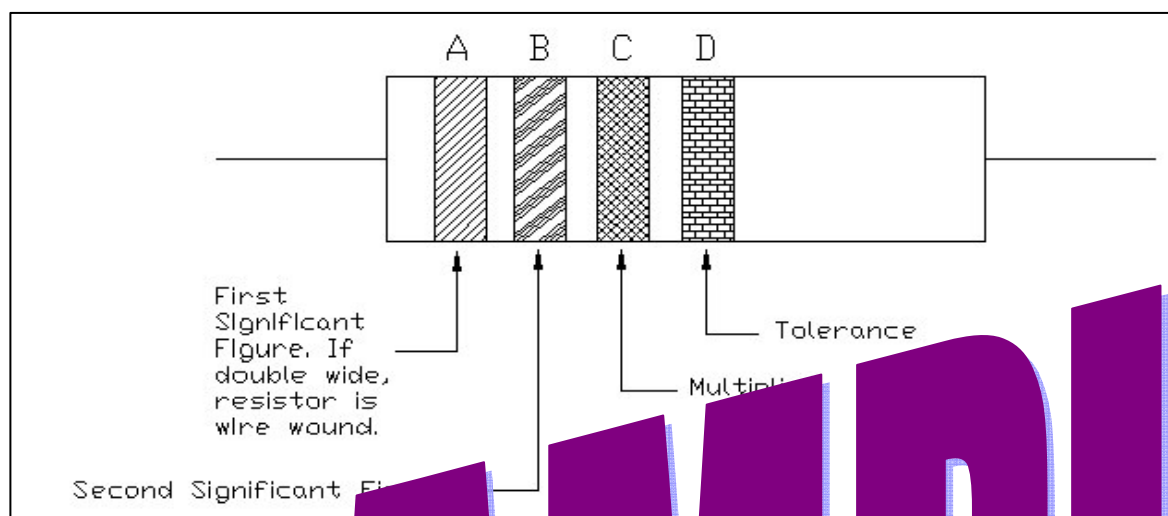
-OR-

$E$	=	10.00 V
$I$	=	20.00 A
$P$	=	200.00 W

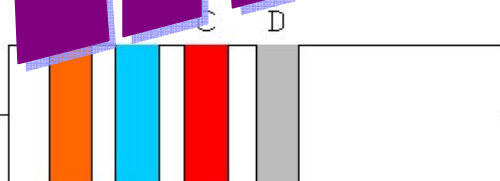
  

$I$	=	20.00 A
$R$	=	0.50 $\Omega$
$P$	=	200.00 W

## Color Code For Resistors



Band	Color	Value	Multiplier	Tolerance
1st	Black	0	1	± 1%
2nd	Brown	1	10	± 1%
3rd	Red	2	100	± 2%
4th	Orange	3	1,000	± 3%
5th	Yellow	4	10,000	± 4%
6th	Green	5	100,000	± 5%
7th	Blue	6	1,000,000	± 0.25%
8th	Purple	7	10,000,000	± 0.1%
9th	Gray	8	100,000,000	± 0.05%
10th	White	9	1,000,000,000	± 0.01%



A = Orange = 3

C = Red = 100

B = Blue = 6

D = Silver = ± 10%

Resistance Value = 3600 Ω ± 10%, or between 3240 Ω and 3960 Ω.

## Resistor Calculator

Band A =	3	Orange
Band B =	6	Blue
Band C =	100	Red
Band D =	± 10 %	Silver

Resistor Value = 3,600 Ω ± 10 %

or between 3,240 Ω and 3,960 Ω

## Reference Electrode Conversion

Relative Values of Typical Reference Electrodes versus Sulfate Reference Electrode

Reference Electrode (E <sub>ref</sub> )	Standard Potential (E <sup>0</sup> )	Conversion Factor
Copper - Sulfate (Cu/CuSO <sub>4</sub> )	+0.337 V	+0.059 V
Silver - Silver Chloride (Ag/AgCl)	+0.222 V	+0.059 V
Saturated Calomel (Hg/Hg <sub>2</sub> Cl <sub>2</sub> )	+0.241 V	+0.059 V
Zinc (Pb/ZnSO <sub>4</sub> )	-0.763 V	-0.059 V

Convert Reference:

Reference Electrode	Standard Potential (E <sup>0</sup> )	Conversion Factor	Reference Electrode
Copper - Sulfate (Cu/CuSO <sub>4</sub> )	+0.337 V	+0.059 V	0.533 V
Silver - Silver Chloride (Ag/AgCl)	+0.222 V	+0.059 V	0.483 V
Saturated Calomel (Hg/Hg <sub>2</sub> Cl <sub>2</sub> )	+0.241 V	+0.059 V	0.463 V
Zinc (Pb/ZnSO <sub>4</sub> )	-0.763 V	-0.059 V	-0.567 V

SAMPLE

## Sacrificial Anode Efficiency and Utilization Factor

Magnesium :

$$L_M = \frac{0.116W}{I}$$

$W$  = Weight of anode ( lb )

$E$  = Current efficiency

$U_F$  = Utilization factor

$I$  = Current ( A )

$L_M$  = Magnesium consumption rate

Z

$W$  = Weight of anode ( lb )

$E$  = Current efficiency

$U_F$  = Utilization factor

$I$  = Current ( A )

$L_Z$  = Zinc consumption rate

$W$

$E$

$U_F$

$I$

$L_Z$

0.90

0.85

2.0 A

3.90 yrs

Anode Material	Consumption Rate <sup>2</sup> ( lb/amp-yr )	Common Consumption Rates Used <sup>2</sup> ( lb/amp-yr )	Current Efficiency <sup>2</sup>	Utilization Factor <sup>3</sup>
Magnesium (H-I Alloy)	15.0 - 35.0	25	0.50	0.85
Magnesium (High Potential)	16.0 - 19.0	17	0.50	0.85
Zinc	23.7	23.7	0.90	0.85

<sup>1</sup>Anodes installed in suitable chemical backfill.

<sup>2</sup>Current efficiency with current density. The shown efficiency, and the resulting consumption rate, are at approximately 30 mA/ft<sup>2</sup> of anode surface. Efficiencies are higher at higher current densities and lower at lower current densities.

<sup>3</sup>The Utilization Factor for all Galvanic Anodes is 85% ( 0.85 ).

# Dwight's Equation for Single Vertical Anode Resistance to Earth - inches

$$R_v = \frac{\rho}{2\pi L} \left\{ \left( \ln \frac{8L}{d} \right) - 1 \right\}$$

$\rho$  = Soil Resistivity ( ohm-cm )

$L$  = Length of Anode ( in )

$d$  = Diameter of Anode ( in )

$R_v$  = Resistance of Single Vertical Anode to Remote Earth ( ohms )

which is

which is

$\rho$	=	10,000 $\Omega$ -cm
$L$	=	472.44 in
$L$	=	1200.00 cm
$d$	=	7.99 in
$d$	=	20.30 cm
$R_v$	=	84 $\Omega$

# Dwight's Equation for Distributed Multiple Vertical Anodes Resistance to Earth - inches

$$R_v = \frac{\rho}{L} \left\{ \left[ \frac{S}{L} \left( \ln \frac{2L}{dS} + \frac{\sqrt{S^2 + L^2}}{L} \right) + \frac{S}{L} - \frac{\sqrt{S^2 + L^2}}{L} - 1 \right] \right\}$$

$\rho$  = Soil Resistivity ( ohm-cm )

$L$  = Length of Anode ( in )

$S$  = Twice the Depth of Anode ( in )

Number of anodes  
Diameter of anode

Resistance of Multiple Vertical Anodes to Remote Earth ( ohms )

which is

$\rho$	=	10,000 $\Omega$ -cm
$L$	=	84.00 in
$L$	=	213.36 cm
$S$	=	118.11 in
$S$	=	300.00 cm
$d$	=	8.00 in
$d$	=	20.30 cm
$R_v$	=	87 $\Omega$

# Dwight's Equation for Distributed Multiple Vertical Anodes Resistance to Earth - inches

$$R_H = \frac{\rho}{L} \left\{ \left[ \frac{S}{L} \left( \ln \frac{2L}{dS} + \frac{\sqrt{S^2 + L^2}}{L} \right) + \frac{S}{L} - \frac{\sqrt{S^2 + L^2}}{L} - 1 \right] \right\}$$

$\rho$  = Soil Resistivity ( ohm-cm )

$L$  = Length of Anode ( in )

$S$  = Twice the Depth of Anode ( in )

$d$  = Diameter of Anode ( in )

$R_H$  = Resistance of Multiple Vertical Anodes to Remote Earth ( ohms )

which is

which is

which is

$\rho$	=	10,000 $\Omega$ -cm
$L$	=	84.00 in
$L$	=	213.36 cm
$S$	=	145.67 in
$S$	=	370.00 cm
$d$	=	8.00 in
$d$	=	20.30 cm
$R_H$	=	87 $\Omega$

# Dwight's Equation for Distributed Multiple Half-Cells

$$R_{\text{total}} = \frac{R}{N} + R_M + R_{\text{anode}}$$

$R$  = Soil Resistivity (ohm-cm)  
 $S$  = Spacing between anodes (m)

$N$  = Number of Anodes

$R_M$  = Resistance of the anode (ohm)

$R_{\text{anode}}$  = Resistance of the anode (ohm)

$R_M$  = Resistance of the anode (ohm) (ohm)

0.001	Ω-cm
0.080	in
0.03	cm
1	nodes
3.00	Ω
3	
0.001	Ω

$R_{Cable}$  = Resistance per Kilometer ( ohm/mi )  
 $L_{Cable}$  = Length of Cable - Sum of Positive and Negative Leads ( ft )  
 $R_C$  = Resistance of Cable ( ohms )

$R_{Cable}$  = 1.341  $\Omega$ /mi  
 $L_{Cable}$  = 383.50 ft  
 $R_C$  = 0.097  $\Omega$

### Rectifier Total Circuit Resistance

$$R_T = R_{Gbed} + R_C + R_S$$

$R_{Gbed}$  = Ground Bed Resistance ( ohms )  
 $R_C$  = Resistance of Cable ( ohms )  
 $R_S$  = Structure/Pipeline-to-Earth Resistance ( ohms )  
 $R_T$  = Total Circuit Resistance ( ohms )

$R_{Gbed}$  = 3.08  $\Omega$   
 $R_C$  = 0.10  $\Omega$   
 $R_S$  =  $\Omega$   
 $R_T$  =  $\Omega$

### Rectifier Efficiency Formula

$$P_{AC} = V_{AC} I_{AC}$$

$$P_{AC}$$

$$\frac{100}{T}$$

$$=$$

$$= \frac{P}{P} = 0$$

$$P_{DC}$$

The Constant Factor is  
 Revolution per second  
 The Observed  
 $P_{AC}$  = Power ( watt )

$$V_{DC}$$

$$A_{DC}$$

$$P_{DC}$$

$$\% \text{ Efficiency}$$

$$\% \text{ Efficiency}$$

= 0.05  
 = 4  
 = 100  
 = 100 W

$$V_{DC} = 2.00 \text{ V}$$

$$A_{DC} = 1.00 \text{ A}$$

$$P_{DC} = 2.00 \text{ W}$$

$$\% \text{ Efficiency} = 27.78 \%$$

### Current Output Calculator

$$I_{Out} = V_{accross_{shunt}} S_{factor}$$

$V_{accross_{shunt}}$  = Voltage Reading Measured Across Shunt ( mV )  
 $S_{factor}$  = Shunt Factor from Chart ( amp/mV )  
 $I_{Out}$  = Current Output of Rectifier ( amps )

$V_{accross_{shunt}}$  = 25 mV  
 $S_{factor}$  = 0.10 A/mV  
 $I_{Out}$  = 2.50 A

## Faraday's Law

$$W_t = KIT$$

$K$  = Consumption Rate - Electrochemical Equivalent ( lb/amp-yr )  
 $I$  = Current ( amps )  
 $T$  = Time ( yrs )  
 $W_t$  = Weight Loss of Product ( lb )

$K$  = 20.10 lb/A-yr  
 $I$  = 0.875 A  
 $T$  = 4.0 yrs  
 $W_t$  = 70.4 lb

Consumption Rate K for Various Metals <sup>1</sup>		
Metal	kg/A-yr	lb/A-yr
Carbon	1.30	2.86
Aluminum	3.00	6.60
Magnesium	4.00	8.80
Iron/Steel	6.30	13.90
High Silicon/Chromium	6.30	13.90
Nickel	6.30	13.90
Monovalent Metals	10.75	23.65
Divalent Metals	21.50	47.30
Trivalent Metals	32.25	70.95

$L_{Desired}$  = Length of Life ( yrs )  
 $I_{Required}$  = Current Required ( amps )  
 $U_{Factor}$  = Utilization Factor  
 $W_{Anode}$  = Weight per Anode ( lb )  
 $\#_{Anodes}$  = # of Anodes to Use

$K$  = 0.75 lb/A-yr  
 $L_{Desired}$  = 20 yrs  
 $I_{Required}$  = 15.00 A  
 $U_{Factor}$  = 0.60  
 $W_{Anode}$  = 60.0 lb  
 $\#_{Anodes}$  = 6.25 anodes  
 $\#_{Anodes}$  = 7 anodes

rounding up - use →

## # of Anodes Required Based on Current Discharge

$$\#_{Anodes} = \frac{I_{Required}}{MD^*}$$

$I_{Required}$  = Current Required ( amps )  
 $MD^*$  = Maximum Discharge per Anode ( amps )  
 $\#_{Anodes}$  = # of Anodes to Use  
 \* From Anode Manufacturer Data

$I_{Required}$  = 14.70 A  
 $MD^*$  = 2.50 A  
 $\#_{Anodes}$  = 5.88 anodes  
 $\#_{Anodes}$  = 6 anodes

rounding up - use →

## Coating Resistance Calculations

$$R_{CE} = A_S R_{P/S}$$

where

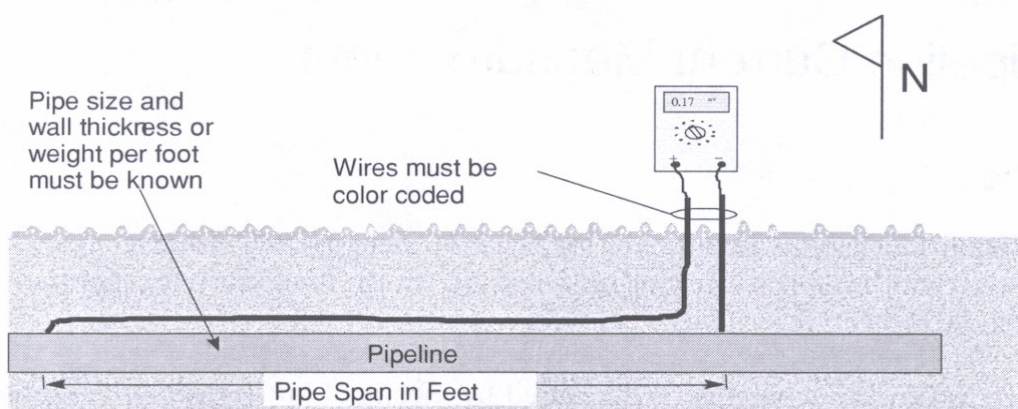
$$A_S = D\pi L$$

$$R_{P/S} = \frac{\Delta E_{ave}}{I_C} = \frac{\frac{\Delta E_{TS1} + \Delta E_{TS2}}{2}}{\frac{\Delta I_{TS1} - \Delta I_{TS2}}{2}} = \frac{(EOFF_{TS1} - EON_{TS1}) - (EOFF_{TS2} - EON_{TS2})}{IOFF_{TS1} - IOFF_{TS2}}$$

Diameter of Pipe (in)  
 Length of Test Span (ft)  
 Area of Pipe Test Span (ft<sup>2</sup>)  
 EON<sub>TS1</sub> = On voltage Reading (Volts)  
 EOFF<sub>TS1</sub> = Off voltage Reading (Volts)  
 ΔE<sub>TS1</sub> = Drop Voltage (Volts)  
 ION<sub>TS1</sub> = Current Reading (amps)  
 IOFF<sub>TS1</sub> = Current Reading (amps)  
 ΔI<sub>TS1</sub> = Current Difference (amps)  
 EON<sub>TS2</sub> = On voltage Reading (Volts)  
 EOFF<sub>TS2</sub> = Off voltage Reading (Volts)  
 ΔE<sub>TS2</sub> = Drop Voltage (Volts)  
 ION<sub>TS2</sub> = Current Reading (amps)  
 IOFF<sub>TS2</sub> = Current Reading (amps)  
 ΔI<sub>TS2</sub> = Current Difference (amps)  
 E<sub>ave</sub> = Average Voltage (volts)  
 I<sub>C</sub> = Current (amps)  
 R<sub>P/S</sub> = Resistance to Soil (ohms)  
 R<sub>CE</sub> = Coating Resistance of Pipe Test Span (ohm-ft<sup>2</sup>)

	in
	50.00 ft
	3.63 ft <sup>2</sup>
EON <sub>TS1</sub>	2.000 V
EOFF <sub>TS1</sub>	0.900 V
ΔE <sub>TS1</sub>	1.100 V
ION <sub>TS1</sub>	2.800 A
IOFF <sub>TS1</sub>	0.100 A
ΔI <sub>TS1</sub>	2.700 A
EON <sub>TS2</sub>	2.000 V
EOFF <sub>TS2</sub>	-0.850 V
ΔE <sub>TS2</sub>	0.850 V
ION <sub>TS2</sub>	2.800 A
IOFF <sub>TS2</sub>	0.100 A
ΔI <sub>TS2</sub>	2.700 A
E <sub>ave</sub>	0.975 V
I <sub>C</sub>	0.100 A
R <sub>P/S</sub>	9.750 Ω
R <sub>CE</sub>	323,296.65 Ω-ft <sup>2</sup>

## 2-Wire Current Test over Pipe Test Span



### 2-wire Line Current Test

Cathodic Protection Level 2 Training Manual - NACE pg. 4:58

$$I_{PS} = \frac{\Delta E_{PS}}{R_{PS}}$$

where

$$R_{PS} = L_{PS} R_P$$

$D$  = Diameter of Pipe ( in )

$t$  = Thickness of Pipe Wall ( in )

$L_{PS}$  = Length of Pipe Test Span ( ft )

$R_P$  = Resistance of Pipe per meter from Table (  $\mu\Omega/\text{ft}$  )

$R_{PS}$  = Resistance of Pipe Test Span (  $\mu\Omega$  )

$\Delta E_{PS}$  = Measured Voltage Drop for Pipe Test Span ( V )

$I_{PS}$  = Calculated Current for Pipe Test Span ( A )

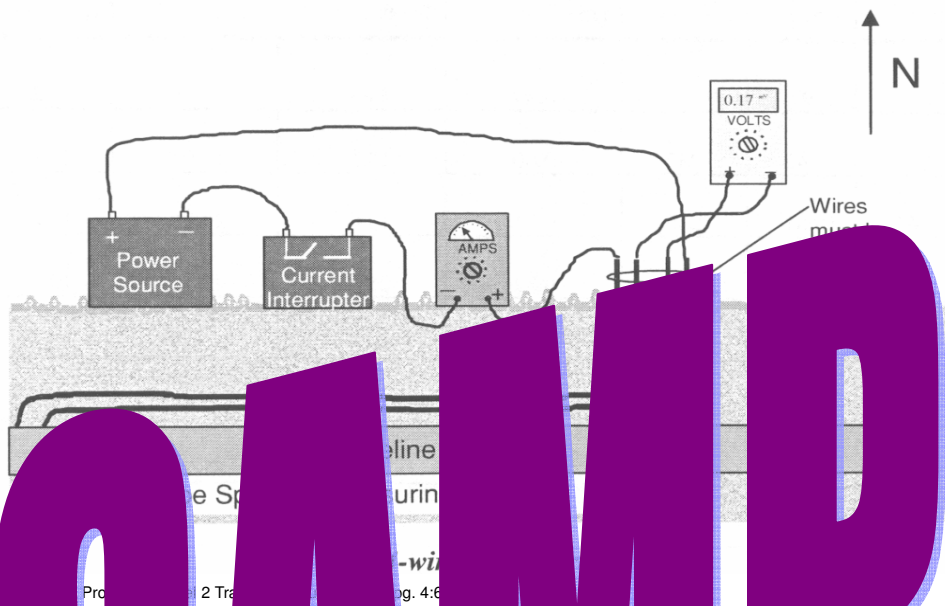
Pipe Size ( in )	Weight ( lb/ft )	Resistance ( $\mu\Omega/\text{ft}$ )	Resistance ( $\mu\Omega$ )
2	3.5	0.92	76.2
4	1.77	0.46	26.8
6	2.60	0.30	15.2
8	3.39	0.20	10.1
10	4.2	0.15	7.1
12	5.0	0.12	5.8
14	5.5	0.10	5.2
16	6.0	0.09	4.6
18	6.5	0.08	4.09
20	7.0	0.07	3.68
22	7.5	0.06	3.34
24	8.0	0.06	3.06
26	8.5	0.05	2.82
28	9.0	0.05	2.62
30	9.5	0.05	2.44
32	10.0	0.05	2.28
34	10.5	0.05	2.15
36	11.0	0.05	2.03

<sup>1</sup> Conversions based on 1in = 2.54cm and 1ft = 0.3048m.

<sup>2</sup> Based on steel density of 489 lbs/ft<sup>3</sup> ( 7832 kg/m<sup>3</sup> ) and steel resistivity of 18  $\mu\Omega/\text{cm}$ .

<sup>3</sup>  $R = 16.061 \times \text{resistivity in } \mu\Omega/\text{cm} = \text{Resistance of 1 ft of Weight per foot of pipe in } \mu\Omega$ .

# 4-Wire Current Test over Pipe Test Span



# SAMPLE

$$I_{PS} = K \frac{E_{ON} - E_{OFF}}{\Delta E_P}$$

$$E_{ON} = \text{Voltage measured across the pipe span when current is applied}$$

$$E_{OFF} = \text{Voltage measured across the pipe span when current is not applied}$$

$E_{ON}$  = Voltage measured across the pipe span when current is applied  
 $E_{OFF}$  = Voltage measured across the pipe span when current is not applied  
 $I_A$  = Applied current (A)  
 $K$  = Calibration factor (A/mV)  
 $\Delta E_P$  = Potential difference between the pipe and the soil (mV)  
 $I_{PS}$  = Current in the pipe (A)

$E_{ON}$	=	0.00017 V
$E_{OFF}$	=	0.00017 V
$I_A$	=	10.00 A
$K$	=	2.037 A/mV
$\Delta E_P$	=	0.00017 V
$I_{PS}$	=	0.346 A

$$C = \frac{1}{R}$$

units are in Siemens ( S ) or microSiemens (  $\mu\text{S}$  )

$$\text{units of Siemen ( S )} = \frac{1}{\Omega}$$

$R$  = Resistance ( ohms )

$C$  = Conductance ( S )

$R$  = 9.750  $\Omega$

$C$  = 0.0103 S

### Resistivity Calculations

- \* Area may be Cross - Sectional Area of Electrode
- \* For Soil Box Measurement & Electrode to Soil Contact Area

$$\frac{RA}{L}$$

Area of Cross Section (  $\text{cm}^2$  )

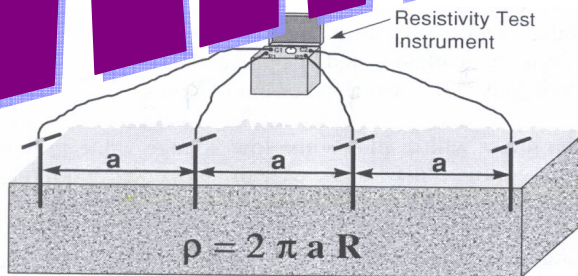
Resistance ( ohms )

$\rho$  = Resistivity ( ohm-cm )

which is

which is

$$\begin{aligned} &= 1.58 \text{ in}^2 \\ &= 10.16 \text{ cm}^2 \\ &= 10.16 \Omega \\ &= 1.58 \text{ in} \\ &= 1.58 \times 2.54 \text{ cm} \\ &= 4.01 \text{ cm} \\ &= 423.332 \Omega\text{-cm} \end{aligned}$$



*Four-Pin (Wenner) Method of Measuring Soil Resistivity*

Cathodic Protection Level 1 Training Manual - NACE 2000 pg 4:28

$$\rho = 2\pi aR$$

$a$  = Spacing of the Pins ( in )

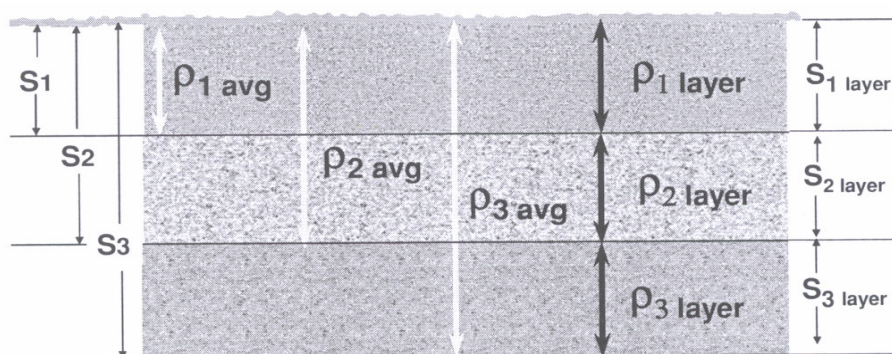
$R$  = Resistance Measured - shown on Instrument ( ohms )

$\rho$  = Resistivity of Measured Medium ( ohm-cm )

which is

$$\begin{aligned} a &= 1.58 \text{ in} \\ a &= 4.00 \text{ cm} \\ R &= 450.00 \Omega \\ \rho &= 11309.734 \Omega\text{-cm} \end{aligned}$$

## Layer Resistivity Calculations



Average and Layer Resistivity

Cathodic Protection Level 2 Training Manual - NACE 2000 pg 4:67

### Parallel Resistance Formula

$$R_c = \frac{R_a R_b}{R_a + R_b}$$

#### Layer 1

$$\rho_{1avg} = 2\pi S_1 R_1$$

$$S_{1layer} = S_1$$

$$R_{1layer} = R_1$$

$$\rho_{1layer} = \rho_1$$

#### Layer 2

$$\rho_{2avg} = 2\pi S_2 R_2$$

$$S_{2layer} = S_2$$

$$R_{2layer} = R_2$$

$$\rho_{2layer} = \rho_2$$

#### Layer 3

$$\rho_{3avg} = 2\pi S_3 R_3$$

$$S_{3layer} = S_3$$

$$R_{3layer} = R_3$$

$$\rho_{3layer} = \rho_3$$

Layer 1 Spacing - Total Depth Measured ( in )	$S_1$	=	62.64 in
Layer 1 Resistance Measured - shown on Instrument ( ohms )	$R_1$	=	7.40 Ω
Layer 1 Average Resistivity ( ohm-cm )	$\rho_{1avg}$	=	9996.90 Ω-cm
Layer 1 Spacing - Layer Depth Measured ( cm )	$S_{1layer}$	=	62.64 in
Layer 1 Parallel Resistance ( ohms )	$R_{1layer}$	=	10.00 Ω
Layer 1 Resistivity ( ohm-cm )	$\rho_{1layer}$	=	9996.90 Ω-cm
Layer 2 Spacing - Total Depth Measured ( in )	$S_2$	=	125.28 in
Layer 2 Resistance Measured - shown on Instrument ( ohms )	$R_2$	=	318.21 cm
Layer 2 Average Resistivity ( ohm-cm )	$\rho_{2avg}$	=	7.40 Ω
Layer 2 Spacing - Layer Depth Measured ( cm )	$S_{2layer}$	=	5824.97 Ω-cm
Layer 2 Parallel Resistance ( ohms )	$R_{2layer}$	=	159.11 cm
Layer 2 Resistivity ( ohm-cm )	$\rho_{2layer}$	=	62.64 in
Layer 3 Spacing - Total Depth Measured ( in )	$S_3$	=	28.46 Ω
Layer 3 Resistance Measured - shown on Instrument ( ohms )	$R_3$	=	28452.714 Ω-cm
Layer 3 Average Resistivity ( ohm-cm )	$\rho_{3avg}$	=	187.92 in
Layer 3 Spacing - Layer Depth Measured ( cm )	$S_{3layer}$	=	477.32 cm
Layer 3 Parallel Resistance ( ohms )	$R_{3layer}$	=	3.10 Ω
Layer 3 Resistivity ( ohm-cm )	$\rho_{3layer}$	=	9297.12 Ω-cm

### Shunt Types and Values

	Shunt Rating		Shunt Value	Shunt Factor
	Amps	mV	Ohms	A/mV
<b>Holloway Type</b>				
RS	5	50	0.01	0.1
SS	25	25	0.001	
SO	50	50	0.01	
SW or CP	1	10	0.05	
SW or CP	2	10	0.025	
SW or CP	3	10	0.017	0.0
SW or CP	4	10	0.025	0.0
SW or CP	5	10	0.01	0.1
SW or CP	10	10	0.05	0.0
SW or CP	15	10	0.033	0.0
SW or CP	20	10	0.025	
SW or CP	25	10	0.02	
SW or CP	30	10	0.017	
SW or CP	40	10	0.01	1
SW or CP	50	10	0.008	1.2
SW or CP	60	10	0.006	1.5
SW or CP	80	10	0.005	2
SW or CP	100	10	0.001	0.1
SW or CP	150	10	0.001	0.01
SW or CP	200	10	0.001	0.01
SW or CP	300	10	0.001	0.1
SW or CP	500	10	0.001	1

$$V_x = \frac{0.038 I \rho}{\pi y} \log_{10} \left( \frac{y + \sqrt{y^2 + x^2}}{x} \right)$$

$\rho$  of Earth  
 $y$  mode

 $x$  is a point in  $A$ 

## Inter-Group Comparisons

$V_x$  = Average Potential (volts) =  $\frac{1}{n} \sum_{i=1}^n V_i$  (volts)

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$E^0$  = Standard Cell Electrode Potential - from table ( volts )  
 $R$  = Gas Constant = 8.31434 ( J/mol·K )  
 $T$  = Absolute Temperature ( K )<sup>1</sup>  
 $n$  = Number of Valence Electrons Transferred - from table  
 $F$  = Faraday's Constant = 96,485.309 ( C/g·mol )  
 $a_M^{+ne}$  = Activity of Metal Ions in Solution<sup>2</sup>  
 $a_M$  = Activity of the Metal (  $a_M = 1$  for pure metal )  
 $E$  = Electrode Potential in Existing Solution ( volts )

$E^0$  = -0.763 V  
 $R$  = 8.31434 J/mol·K  
 $T$  = 298.15 K  
 $n$  = 2  
 $F$  = 96485.31 C/g·mol  
 $a_M^{+ne}$  = 0.71  
 $a_M$  = 1  
 $E$  = -0.997 V

<sup>1</sup> Absolute Temperature is -273.15 °C. Standard Conditions for pure metals in the tables are based on one unit activity of metal ions in the solution at 25 °C with no pressure in the metal or electrolyte and with the standard hydrogen electrode.

<sup>2</sup> Activity Coefficients are calculated using the Debye-Huckel equation at a concentration of 0.01 mol/L.

Periodic Table of Elements

Half-Cell	Standard Potential (Volts)
Au/Au <sup>+</sup>	-1.50 to -1.7
Pt/Pt <sup>++</sup>	-1.10
Cu/Cu <sup>+</sup>	-1.05
H <sub>2</sub> /2H <sup>+</sup>	-0.45 to -0.5
Pb/Pb <sup>++</sup>	-0.15
Ni/Ni <sup>++</sup>	-0.90
Fe/Fe <sup>++</sup>	-0.90
Zn/Zn <sup>++</sup>	-0.90
Al/Al <sup>++</sup>	-0.90
Mg/Mg <sup>++</sup>	-0.90

\* Standard conditions are 25 °C and 1 atm.

